

*Biochimica et Biophysica Acta*, 569 (1979) 31–40  
 © Elsevier/North-Holland Biomedical Press

BBA 68773

## THE SUSCEPTIBILITY OF UROKINASE TO AFFINITY LABELING BY PEPTIDES OF ARGININE CHLOROMETHYL KETONE

CHARLES KETTNER and ELLIOTT SHAW

*Biology Department, Brookhaven National Laboratory, Upton, NY 11973 (U.S.A.)*

(Received November 7th, 1978)

*Key words: Urokinase, Plasminogen activator; Serine protease; Arginine chloromethyl ketone*

### Summary

Pro-Gly-ArgCH<sub>2</sub>Cl, a reagent corresponding to the C-terminal sequence generated in plasminogen on activation by urokinase (EC 3.4.99.26) and probably by other plasminogen activators, was prepared. Pro-Gly-ArgCH<sub>2</sub>Cl was effective in the inactivation of urokinase at the 10<sup>-6</sup> M level ( $K_i$  68  $\mu$ M and  $k_2$  0.47 min<sup>-1</sup>). In contrast, only a slow inactivation was obtained by 10<sup>-2</sup> M *N*-tosyllysine chloromethyl ketone. Glu-Gly-ArgCH<sub>2</sub>Cl, *N,N*-dimethylaminonaphthalene-5-sulfonyl-Glu-Gly-ArgCH<sub>2</sub>Cl, and Ac-Gly-Gly-ArgCH<sub>2</sub>Cl were more reactive than Pro-Gly-ArgCH<sub>2</sub>Cl against urokinase by factors of 25, 6, and 3, respectively. The effectiveness of arginine chloromethyl ketones as affinity labels is highly dependent on binding in the S<sub>2</sub> and S<sub>3</sub> sites, thus sequence variations in the reagents exhibited differences in reactivity of up to four orders of magnitude. The most effective reagents had Gly in P<sub>2</sub>.

Ac-Gly-Gly-ArgCH<sub>2</sub>Cl inactivates urokinase 50 times more rapidly than it does plasmin, thus providing a means of distinguishing the activity of plasmin from its activating protease whereas urokinase is almost inert to Ala-Phe-LysCH<sub>2</sub>Cl, a reagent which inactivates plasmin at the 10<sup>-7</sup> M level.

### Introduction

Tissue and cellular plasminogen activators are serine proteases associated with a number of pathological processes which include malignant cell transfor-

---

Abbreviations: Z, carbobenzoxy; Boc, *t*-butoxycarbonyl; DNS, *N,N*-dimethylaminonaphthalene-5-sulfonyl; OBzl, benzyl ester; Pipes, piperazine-*N,N'*-bis(2-ethanesulfonic acid); TLC, thin-layer chromatography.

mation [1,2], inflammation [3], and the promotion of carcinogenesis [4]. The association of plasminogen activators with these processes is marked by notable increases in plasminogen-dependent fibrinolytic activity although the exact contribution of the proteases to the pathological process is not certain. On the other hand, plasma plasminogen activators have an essential role in haemostasis. The formation of plasminogen activator activity is associated with the initial stages of the Hageman factor-mediated blood coagulation pathway, but the identity of the protease remains obscure [5–7]. Clearly, specific inhibitors for plasminogen activators would be helpful in clarifying the physiological roles of these proteases and in establishing their identity.

An approach which has yielding high specific irreversible inhibitors for other physiologically important trypsin-like proteases such as thrombin [8], plasma kallikrein [9] and factor Xa (unpublished data) has been the synthesis of chloromethyl ketones which contain the amino acid sequence at the cleavage site of the physiological substrate for the target protease. Although these proteases have a common primary specificity favoring arginine, secondary sites are sufficiently different to permit such reagents to act selectively.

The urinary plasminogen activator, urokinase (EC 3.4.99.26), was chosen for initial studies and an affinity label was prepared which contains the Pro-Gly-Arg- sequence of plasminogen hydrolyzed by urokinase in the activation of the zymogen [10]. The reactivity of the reagent corresponding to this amino acid sequence, Pro-Gly-ArgCH<sub>2</sub>Cl, was compared with that of other arginine chloromethyl ketones in the inactivation of urokinase in order to obtain an insight to the specificity of urokinase. In addition, the selectivity of affinity labels for urokinase was evaluated by comparing their reactivity with the plasminogen activator with that for certain other physiologically important proteases.

## Materials and Methods

*Analytical procedures.* Elemental microanalyses were performed by Galbraith Laboratories, Inc., Knoxville, TN. Single column amino acid analyses were performed on a 0.9 × 22 cm column packed with Durrum DC-6A resin on samples hydrolyzed in 2 ml of 6.0 N HCl for 24 h at 110°C in sealed evacuated tubes. TLC was performed on E. Merck precoated silica gel plates (No. 5534) using butanol/acetic acid/water (4 : 1 : 1) as a developing solvent.

*Preparation of Pro-Gly-ArgCH<sub>2</sub>Cl · 2HCl.* H-Arg(NO<sub>2</sub>)CH<sub>2</sub>Cl · HCl (0.49 g, 1.7 mmol) was prepared and was coupled, using the mixed anhydride method, to Z-Pro-Gly-OH (0.59 g, 1.9 mmol) by the procedures described previously [9]. After removing the reaction solvent, the residue was dissolved in ethylacetate and washed with 0.2 M HCl, 5% NaHCO<sub>3</sub>, and saturated aqueous NaCl. The crude product was purified by chromatography on a silica gel column packed with 5 μm LiChrosorb SI 60 (E. Merck, 5945143) using ethylacetate containing 3% methanol as a developing solvent. After evaporating the solvent and triturating with ether, 0.34 g of Z-Pro-Gly-Arg(NO<sub>2</sub>)CH<sub>2</sub>Cl were obtained.

Analysis: for C<sub>22</sub>H<sub>30</sub>N<sub>7</sub>O<sub>7</sub>Cl: Calcd.: C = 48.93%, H = 5.61%, N = 18.16%.

Found: C = 48.62%, H = 5.89%, N = 17.88%.

Z-Pro-Gly-Arg(NO<sub>2</sub>)CH<sub>2</sub>Cl (0.20 g, 0.37 mmol) was treated with approximately 10 ml of anhydrous hydrogen fluoride in the presence of 1 ml of anisole for 30 min at 0°C. HF was removed by evaporation, the residue was taken up in 20 ml of water and extracted with two 20-ml portions of ether. The aqueous phase was applied to a column containing 10 ml of SE-Sephadex (C-25, H<sup>+</sup>) which was washed with 100 ml of water. The product was eluted with 120 ml of 0.4 N HCl. After lyophilization and trituration with ether, 110 mg of Pro-Gly-ArgCH<sub>2</sub>Cl · 2HCl were obtained.

TLC indicated a single spot at  $R_F$  0.074 by both ninhydrin and Sakaguchi stains. Amino acid analysis of Pro-Gly-ArgCH<sub>2</sub>Cl · 2HCl: Pro, 1.00 and Gly, 0.94.

*Preparation of Phe-Ala-ArgCH<sub>2</sub>Cl · 2HCl.* Boc-Phe-Ala-Arg(NO<sub>2</sub>)CH<sub>2</sub>Cl was prepared by coupling Boc-Phe-Ala-OH (0.58 g, 1.7 mmol) to H-Arg(NO<sub>2</sub>)CH<sub>2</sub>Cl · HCl (0.05 g, 1.7 mmol) by the mixed anhydride procedure described previously [9]. After crystallization from ethylacetate, 0.57 g of product (m.p. 110–114°C) were obtained.

Analysis: for C<sub>24</sub>H<sub>36</sub>N<sub>7</sub>O<sub>7</sub>Cl: Calcd.: C = 50.56%, H = 6.38%, N = 17.20%.

Found: C = 50.75%, H = 6.45%, N = 16.94%.

Boc-Phe-Ala-Arg(NO<sub>2</sub>)CH<sub>2</sub>Cl (0.40 g, 0.70 mmol) was deblocked with HF and the product was isolated by the procedure described for the preparation of Pro-Gly-ArgCH<sub>2</sub>Cl · 2HCl to yield 0.28 g of Phe-Ala-ArgCH<sub>2</sub>Cl · 2HCl. TLC indicated a single spot,  $R_F$  0.47, by both ninhydrin and Sakaguchi stains. Amino acid analysis of Phe-Ala-ArgCH<sub>2</sub>Cl · 2HCl: Phe, 0.97 and Ala, 1.00.

*Preparation of Glu-Gly-ArgCH<sub>2</sub>Cl · 2 HCl* Boc-Glu(OBzl)-Gly-Arg(NO<sub>2</sub>)CH<sub>2</sub>Cl was prepared by coupling Boc-Glu(OBzl)-Gly-OH (0.68 g, 1.7 mmol) to H-Arg(NO<sub>2</sub>)CH<sub>2</sub>Cl · HCl (0.50 g, 1.7 mmol) by the mixed anhydride procedure described previously [9]. Boc-Glu(OBzl)-Gly-Arg(NO<sub>2</sub>)CH<sub>2</sub>Cl was crystallized from ethylacetate/hexane to yield 0.56 g.

Analysis: for C<sub>26</sub>H<sub>38</sub>N<sub>7</sub>O<sub>9</sub>Cl: Calcd.: C = 49.71%, H = 6.11%, N = 15.61%.

Found: C = 49.69%, H = 6.26%, N = 15.68%.

Boc-Glu(OBzl)-Gly-Arg(NO<sub>2</sub>)CH<sub>2</sub>Cl (0.30 g, 0.48 mmol) was reacted with HF for 70 min at 0°C followed by isolation as described in the preparation of Pro-Gly-ArgCH<sub>2</sub>Cl · 2HCl to yield 0.18 g of Glu-Gly-ArgCH<sub>2</sub>Cl · 2HCl. TLC indicated a major spot at  $R_F$  0.14 and a trace at  $R_F$  0.31 by both ninhydrin and Sakaguchi stains. Amino acid analysis of Glu-Gly-ArgCH<sub>2</sub>Cl · 2HCl: Glu, 1.00 and Gly, 1.00.

*Preparation of DNS-Glu-Gly-ArgCH<sub>2</sub>Cl · HCl.* H-Glu(OBzl)-Gly-Arg(NO<sub>2</sub>)-CH<sub>2</sub>Cl · HCl was prepared by treating Boc-Glu(OBzl)-Gly-Arg(NO<sub>2</sub>)CH<sub>2</sub>Cl with 2 ml of anhydrous trifluoroacetic acid for 5 min at room temperature followed by 0.5 ml of ethanolic HCl (5 N) and precipitation with cold ether.

H-Glu(OBzl)-Gly-Arg(NO<sub>2</sub>)CH<sub>2</sub>Cl · HCl (0.50 g, 0.87 mmol) was dissolved in 4 ml of water and 2 ml of dioxane and cooled to 0°C. After addition of NaHCO<sub>3</sub> (0.19 g, 2.2 mmol) and dansyl chloride (0.31 g, 1.2 mmol) dissolved in 4 ml of dioxane, the solution was stirred for 2.5 h at room temperature. The

reaction solution was added to 100 ml of ethylacetate and was washed with 5%  $\text{NaHCO}_3$  and saturated aqueous  $\text{NaCl}$ . The product was purified by chromatography on a column packed with silica gel 60, 40–63  $\mu\text{m}$  particle size (E. Merck, 9385) using chloroform/methanol (95 : 5) as a developing solvent. After evaporating solvent and triturating with ether, 0.31 g of  $\text{DNS-Glu(OBzl)-Gly-Arg(NO}_2\text{)CH}_2\text{Cl}$  were obtained.

Analysis: for  $\text{C}_{33}\text{H}_{41}\text{N}_8\text{O}_9\text{SCl}$ : Calcd.: C = 52.06%, H = 5.44%, N = 14.72%.

Found: C = 52.03%, H = 5.47%, N = 14.92%.

$\text{DNS-Glu(OBzl)-Gly-Arg(NO}_2\text{)CH}_2\text{Cl}$  (0.29, 0.38 mmol) was reacted with HF for 60 min at  $0^\circ\text{C}$  and the product was isolated by the procedure described in the preparation of  $\text{Pro-Gly-ArgCH}_2\text{Cl} \cdot 2\text{HCl}$  to yield 0.23 g of  $\text{DNS-Glu-Gly-ArgCH}_2\text{Cl} \cdot \text{HCl}$ . TLC indicated a single spot,  $R_F$  0.47, by Sakaguchi stain. The plate was ninhydrin negative.

*Preparation of Ac-Gly-Gly-ArgCH<sub>2</sub>Cl · HCl.*  $\text{Ac-Gly-Gly-OH}$  (0.25 g, 0.88 mmol) was coupled to  $\text{H-Arg(NO}_2\text{)CH}_2\text{Cl} \cdot \text{HCl}$  (0.25 g, 0.88 mmol) by the procedure described previously [9] except dimethylformamide was used as a solvent for the formation of the mixed anhydride. After removing the reaction solvent, the residue was crystallized from methanol to yield 0.19 g of  $\text{Ac-Gly-Gly-Arg(NO}_2\text{)CH}_2\text{Cl}$ .

Analysis: for  $\text{C}_{13}\text{H}_{22}\text{N}_7\text{O}_6\text{Cl}$ : Calcd.: C = 38.28%, H = 5.45%, N = 24.04%.

Found: C = 37.90%, H = 5.07%, N = 23.88%.

$\text{Ac-Gly-Gly-ArgCH}_2\text{Cl} \cdot \text{HCl}$  was prepared by deblocking  $\text{Ac-Gly-Gly-Arg(NO}_2\text{)CH}_2\text{Cl}$  (0.29 g, 0.71 mmol) with HF by the procedure described for the preparation of  $\text{Pro-Gly-ArgCH}_2\text{Cl} \cdot 2\text{HCl}$  to yield 0.18 g of product. TLC indicated a single spot,  $R_F$  0.29, by Sakaguchi stain. The plate was ninhydrin negative.

*Enzymatic studies.* Human urokinase was obtained from Leo Pharmaceutical Products, Denmark; No. 9092207 1712 E. Stock solutions consisting of 1990 Ploug units or 1.1 nmol of the protease established as described in [11] in 2.0 ml of 50 mM Pipes buffer, pH 6.0, which was 0.20 M in  $\text{NaCl}$  and of the affinity labeling reagent in 1.0 mM  $\text{HCl}$  were prepared. Aliquots of the affinity labeling reagent, not exceeding 0.200 ml, were diluted to the desired concentration by the addition of 50 mM Pipes buffer, pH 7.0, which was 0.20 M in  $\text{NaCl}$  to yield a final volume of 2.00 ml. Reactions were initiated by the addition of 50  $\mu\text{l}$  of the stock urokinase to the inhibitor solution at  $25^\circ\text{C}$ . At least eight, timed aliquots (200  $\mu\text{l}$ ) were removed and residual esterase activity was determined using the thiobenzyl Z-lysinate assay [12]. The apparent, pseudo first-order rate constants for the inactivation of urokinase were determined from the slope of semilogarithmic plots of esterase activity vs. time.

Kinetic studies were based on the affinity labeling mechanism for serine proteases, Eqn. 1 [13]. Values of  $K_1$ , the reversible dissociation constant for the protease-inhibitor complex, and of  $k_2$ , the first-order rate constant for the alkylation of the active-site histidine, were determined by the graphic solution of Eqn. 2 [14] where  $k_{app}$  is the apparent, pseudo first-order rate constant and

$I$  is the concentration of the affinity label. At least seven values



$$\frac{1}{k_{app}} = \frac{K_i}{k_2} \frac{1}{I} + \frac{1}{k_2} \quad (2)$$

of  $k_{app}$  were determined at different concentrations of the affinity labels and kinetic constants were determined from double-reciprocal plots of  $k_{app}$  vs.  $I$  after determining the best straight line by the least squares method.

Additional kinetic constants,  $k_{app}/I$ , were determined for human urokinase, plasma kallikrein, plasmin, and bovine thrombin by a procedure we have previously described [9]. Values of  $k_{app}/I$  for purified human urinary kallikrein (17 tosylarginine methyl ester units/mg), a gift from Dr. Jack V. Pierce, were measured by the procedure described for urokinase except  $2.5 \cdot 10^{-3}$  unit [15] were incubated with the affinity labels and the protease was assayed using 0.30 mM substrate. Values of  $k_{app}$  were measured at several concentrations of the affinity label to establish that this constant was proportional to  $I$  in the range studied. The constant,  $k_{app}/I$ , was then used as an estimate of  $k_2/K_i$  according to the relationship of Eqn. 3 [14].

$$\frac{k_{app}}{I} = \frac{k_2}{K_i} \quad \text{if } I \ll K_i \quad (3)$$

## Results

Syntheses were developed for Pro-Gly-ArgCH<sub>2</sub>Cl and a number of closely related peptides containing arginine chloromethyl ketone which were of interest in exploring the relation of structure to ability to inactivate urokinase. Pro-Gly-Arg is the sequence in plasminogen at which cleavage of the arginyl peptide bond results in activation [10,16]. The effectiveness of the newly synthesized compounds as affinity labels for urokinase was compared with that of a group containing sequences specific for other proteases.

The rates of inactivation of urokinase were measured at concentrations of reagents so that the half-times for inactivation were in the range of 20–30 min. This rate of inactivation is favorable for accurate rate measurement and was generally obtainable with concentrations of affinity labeling reagent for which the rates of inactivation were proportional to the concentration used, therefore below a level producing saturation effects.

Pro-Gly-ArgCH<sub>2</sub>Cl was in fact much more effective in inactivating urokinase than many tripeptides containing arginine chloromethyl ketone or lysine chloromethyl ketone. It was superior to Tos-LysCH<sub>2</sub>Cl, a commonly used reagent for the inactivation of trypsin-like enzymes, by four orders of magnitude (Table I). However, a number of reagents having in common a glycine residue in the P<sub>2</sub> position were even more effective (Fig. 1). Thus, Glu-Gly-ArgCH<sub>2</sub>Cl inactivated urokinase at least 20 times more rapidly than Pro-Gly-ArgCH<sub>2</sub>Cl as shown by the values of  $k_{app}/I$  in Table I. DNS-Glu-Gly-ArgCH<sub>2</sub>Cl and Ac-Gly-Gly-ArgCH<sub>2</sub>Cl were less effective, differing in reactivity from Glu-Gly-ArgCH<sub>2</sub>Cl by factors of 4 and 8, respectively.

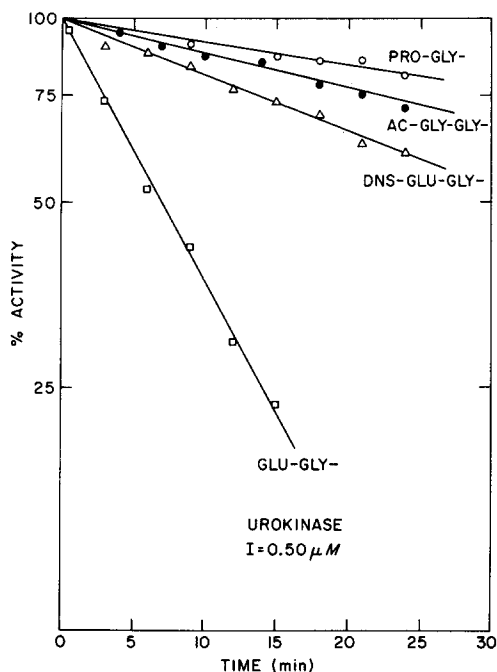


Fig. 1. Comparison of the reactivity of urokinase with different N-substituted arginine chloromethyl ketones. The reagents were incubated at a concentration of  $0.50 \mu\text{M}$  with urokinase at pH 7.0 and  $25^\circ\text{C}$ . The inactivation of urokinase was monitored by removing timed aliquots and assaying for esterase activity.

Pro-Gly-ArgCH<sub>2</sub>Cl, however, was more effective than arginine chloromethyl ketones which contained residues other than Gly in the P<sub>2</sub> position. The role of binding in secondary sites in determining the selectivity of urokinase is apparent in the large differences observed among arginine chloromethyl ketones with different substituents in the P<sub>2</sub> and P<sub>3</sub> positions. For example, urokinase differed in its reactivity with Pro-Phe-ArgCH<sub>2</sub>Cl and Glu-Gly-ArgCH<sub>2</sub>Cl by four orders of magnitude. On the other hand, the contribution of arginine in the P<sub>1</sub> position is illustrated by the ten-fold greater reactivity of Phe-Ala-ArgCH<sub>2</sub>Cl compared to its lysine analog.

The difference in the effectiveness of Glu-Gly-ArgCH<sub>2</sub>Cl and the other reagents in Table I in the inactivation of urokinase is probably due to differences in affinity since analysis of the reactivity of peptides of arginine chloromethyl ketone with other proteases indicated that the rate constants for the alkylation step of the affinity labeling mechanism,  $k_2$ , often were relatively constant for individual proteases [8,9]. This has been found to be the case for the 26-fold difference in the reactivities of Glu-Gly-ArgCH<sub>2</sub>Cl and Pro-Gly-ArgCH<sub>2</sub>Cl with urokinase after evaluating values of  $K_1$  and  $k_2$  for these reagents by the method of Kitz and Wilson [14]. The values of  $k_2$  for the inactivation of urokinase by Glu-Gly-ArgCH<sub>2</sub>Cl and Pro-Gly-ArgCH<sub>2</sub>Cl were almost identical,  $0.46 \text{ min}^{-1}$  and  $0.47 \text{ min}^{-1}$ , respectively, whereas differences in the dissociation constant,  $K_1$ , for reversible binding of the two reagents were comparable to differences in their reactivity. The  $K_1$  for Glu-Gly-ArgCH<sub>2</sub>Cl was  $1.8 \cdot 10^{-6} \text{ M}$

TABLE I

## COMPARISON OF THE REACTIVITY OF UROKINASE WITH PEPTIDES OF ARGININE AND LYSINE CHLOROMETHYL KETONES

Inactivations were conducted at 25°C in 50 mM Pipes buffer, pH 7.0, containing 0.20 M NaCl. The initial concentration of urokinase was 13 nM.  $t_{1/2}$  is the half-time for the first-order inactivation of urokinase at the indicated concentration of the affinity labeling reagent.  $k_{app}/I$  is the ratio of the apparent, first-order rate constant for inactivation to the concentration of the affinity label. These values were calculated from the values of  $t_{1/2}$  using the relationship,  $k_{app} = \ln 2/t_{1/2}$ , and the indicated concentration of the affinity label.

Affinity label	Concn. ( $\mu$ M)	$t_{1/2}$ (min)	$k_{app}/I$ ( $M^{-1} \cdot \text{min}^{-1}$ ) ( $\times 10^{-4}$ )
Glu-Gly-ArgCH <sub>2</sub> Cl	0.20	17.0	20.4
DNS-Glu-Gly-ArgCH <sub>2</sub> Cl	0.50	28.9	4.80
Ac-Gly-Gly-ArgCH <sub>2</sub> Cl	0.50	54.0	2.57
Pro-Gly-ArgCH <sub>2</sub> Cl	2.0	44.0	0.79
Val-Pro-ArgCH <sub>2</sub> Cl	10.0	12.8	0.54
Ile-Pro-ArgCH <sub>2</sub> Cl	10.0	17.8	0.39
Phe-Ala-ArgCH <sub>2</sub> Cl	10.0	23.9	0.29
Val-Ile-Pro-ArgCH <sub>2</sub> Cl	20.0	19.6	0.18
Phe-Ala-LysCH <sub>2</sub> Cl	100.0	28.7	0.024
Gly-Val-ArgCH <sub>2</sub> Cl	100.0	41.3	0.017
Ile-Leu-ArgCH <sub>2</sub> Cl	200.0	23.9	0.014
Ala-Phe-ArgCH <sub>2</sub> Cl	500.0	23.6	0.0059
Pro-Phe-ArgCH <sub>2</sub> Cl	500.0	91	0.0015
Ala-Phe-LysCH <sub>2</sub> Cl	2 500	83	0.0003
Tos-LysCH <sub>2</sub> Cl	50 000	85	0.00002

and for Pro-Gly-ArgCH<sub>2</sub>Cl was  $6.9 \cdot 10^{-5}$  M.

To determine to what extent the reagents effective in inactivating urokinase are selective, the reactivities of urokinase with Glu-Gly-ArgCH<sub>2</sub>Cl, DNS-Glu-Gly-ArgCH<sub>2</sub>Cl, Ac-Gly-Gly-ArgCH<sub>2</sub>Cl, and Pro-Gly-ArgCH<sub>2</sub>Cl were compared with the reactivities of the plasma proteases, plasma kallikrein, thrombin, and plasmin. All reagents in Table II except for DNS-Glu-Gly-ArgCH<sub>2</sub>Cl inactivated urokinase more effectively than plasmin as shown by the differences in the second-order constant for their inactivation reactions. For example, Ac-Gly-Gly-ArgCH<sub>2</sub>Cl inactivated plasmin by 50% in 70 min at a concentration

TABLE II

## SELECTIVITY OF ARGININE CHLOROMETHYL KETONES IN THE INACTIVATION OF TRYPSIN-LIKE PROTEASES

$k_{app}/I$  is the ratio of the apparent, first-order rate constant for inactivation to the concentration of the affinity label. Values of  $k_{app}$  were measured at 25°C, pH 7.0.

Affinity label	$k_{app}/I$ ( $M^{-1} \cdot \text{min}^{-1}$ ) ( $\times 10^{-4}$ )				
	Uro- kinase	Plasma kallikrein	Thrombin	Plasmin	Urinary kallikrein
Glu-Gly-ArgCH <sub>2</sub> Cl	20	16	1.9	1.3	0.00095
DNS-Glu-Gly-ArgCH <sub>2</sub> Cl	4.2	140	26	28	0.18
Ac-Gly-Gly-ArgCH <sub>2</sub> Cl	2.6	1.4	0.74	0.053	0.00054
Pro-Gly-ArgCH <sub>2</sub> Cl	0.79	3.3	1.2	0.091	0.0017

of  $2.0 \cdot 10^{-5}$  M while urokinase was inactivated by 50% in 54 min at a concentration of  $5.0 \cdot 10^{-7}$  M. Plasma kallikrein, thrombin, and urokinase were similar in their reactivities with most of the reagents in Table II while in contrast human urinary kallikrein was almost inert to inactivation by Glu-Gly-Arg-CH<sub>2</sub>Cl, Ac-Gly-Gly-ArgCH<sub>2</sub>Cl and Pro-Gly-ArgCH<sub>2</sub>Cl, requiring concentrations of reagents 3–5 orders of magnitude greater than effective with urokinase. A unique difference in the reactivity of urokinase and the other proteases is the 9–20-fold greater reactivities of the plasma proteases and the 180-fold greater reactivity of urinary kallikrein with DNS-Glu-Gly-ArgCH<sub>2</sub>Cl than with Glu-Gly-ArgCH<sub>2</sub>Cl while the bulky DNS residue in the P<sub>4</sub> position was restrictive for urokinase, decreasing the effectiveness of the reagent four-fold.

## Discussion

Urokinase has been the object of a number of studies involving synthetic inhibitors acting reversibly [17,18], or irreversibly by acylation [19], sulfonylation [20], or alkylation [21]. The present work extends to this enzyme alkylation by peptides of arginine chloromethyl ketone which have proven to be very effective with other trypsin-like enzymes [8,9,22].

The preparation of Pro-Gly-ArgCH<sub>2</sub>Cl, an affinity label corresponding to the amino acid sequence of plasminogen cleaved by urokinase in zymogen activation, yielded a reagent effective in the inactivation of urokinase, but lacking the high degree of reactivity and selectivity of reagents for thrombin [8] and plasma kallikrein [9] corresponding to their respective physiological substrates. In fact, Pro-Gly-ArgCH<sub>2</sub>Cl was more effective in the inactivation of plasma kallikrein and thrombin than in the inactivation of urokinase.

Other reagents, Glu-Gly-ArgCH<sub>2</sub>Cl, DNS-Glu-Gly-ArgCH<sub>2</sub>Cl, and Ac-Gly-Gly-ArgCH<sub>2</sub>Cl, proved to be more effective than Pro-Gly-ArgCH<sub>2</sub>Cl in the inactivation of urokinase. Substitution of Glu for Pro in the P<sub>3</sub> position of Pro-Gly-ArgCH<sub>2</sub>Cl enhanced the affinity of urokinase for the reagent 40-fold while a three-fold enhancement was obtained by substitution of Ac-Gly- for Pro in the P<sub>3</sub> position. In contrast to the P<sub>3</sub> position, the importance of the P<sub>2</sub> glycyl residue in determining the selectivity of urokinase for the single -Arg-Val- bond of the -Pro-Gly-Arg-Val-Val- sequence of plasminogen [10] is clearly demonstrated. Bulkier substituents in the P<sub>2</sub> position were progressively less effective inactivators of urokinase. In all cases, tripeptide analogs were more effective than the corresponding tetrapeptide analogs. For example, Glu-Gly-ArgCH<sub>2</sub>Cl is four times more effective than DNS-Glu-Gly-ArgCH<sub>2</sub>Cl and Ile-Pro-ArgCH<sub>2</sub>Cl is twice as effective as Val-Ile-Pro-ArgCH<sub>2</sub>Cl. An additional example is the 68-fold greater reactivity of Glu-Gly-ArgCH<sub>2</sub>Cl than Ile-Glu-Gly-ArgCH<sub>2</sub>Cl with urokinase (unpublished observation). The restrictive nature of the S<sub>4</sub> subsite of urokinase appears to be a feature distinguishing urokinase from the plasma proteases examined. For the latter, tetrapeptide analogs were similar in reactivity to the corresponding tripeptide analogs and in several cases enhanced reactivities, similar to that found with DNS-Glu-Gly-ArgCH<sub>2</sub>Cl, were observed.

The lower reactivity of urokinase with the reagent corresponding to the amino acid sequence of its physiological substrate than that expected from the reactivities of the plasma proteases, thrombin [8] and plasma kallikrein [9],



with their appropriate affinity labels was not due to differences in the susceptibility of the active-site histidine to alkylation since the first-order rate constant for the alkylation step of the affinity labeling reaction was comparable to those measured for thrombin [8], plasmin, and plasma kallikrein [9]. The lower reactivity of urokinase most probably reflects its high degree of specificity. A much larger binding region of a unique secondary structure of the polypeptide chain at the urokinase cleavage site of plasminogen, which is not obtained with the tripeptide analogs, may be essential in determining the selectivity of urokinase. The relatively low reactivity of urokinase with Pro-Gly-ArgCH<sub>2</sub>Cl is consistent with the observation of Sottrup-Jensen et al. [10] that urokinase does not hydrolyze a 38 residue, carboxymethylated fragment of plasminogen encompassing the region cleaved by urokinase in the activation of plasminogen. Also consistent with the high degree of specificity of urokinase, large differences in the reactivity of urokinase with arginine chloromethyl ketones containing different substituents in the P<sub>2</sub> and P<sub>3</sub> binding sites were observed. The difference in the reactivity of urokinase with the most effective and least effective arginine chloromethyl ketone spans four orders of magnitude. A similar comparison for thrombin [8] revealed differences of three orders in magnitude while for the less specific proteases, trypsin and acrosin, differences of 180 and 100 were obtained [22].

A selective affinity label for urokinase has not emerged from these initial studies in which the reactivity of the urinary plasminogen activator was compared with the reactivities of plasma proteases. However, in contrast to the similar reactivities of plasma kallikrein and urokinase with Glu-Gly-ArgCH<sub>2</sub>Cl, Ac-Gly-Gly-ArgCH<sub>2</sub>Cl, and Pro-Gly-ArgCH<sub>2</sub>Cl, human urinary kallikrein was decisively insensitive to these reagents thus providing a means for distinguishing between the two urinary proteases. Powerful reagents are now available which may be useful in determining the roles of urokinase and other plasminogen activators in their physiological environments. For example, Ac-Gly-Gly-Arg-CH<sub>2</sub>Cl is 50 times more reactive with urokinase than with plasmin, thus providing a means of distinguishing the activity of plasmin from its activating protease. Other affinity labels readily inactivate plasmin distinguishing it from urokinase. The most effective, Ala-Phe-LysCH<sub>2</sub>Cl [9], inactivates plasmin at the 10<sup>-7</sup> M level while urokinase only slowly reacts with this reagent even at 10<sup>-3</sup> M.

In addition, the large differences in reactivity of urokinase with various arginine chloromethyl ketones has made it possible to distinguish urokinase from the plasminogen activator secreted by HeLa cells [23]. The series of arginine chloromethyl ketones described here should be useful in characterizing plasminogen activators from other sources [23] and may lead to reagents of therapeutic value.

#### Acknowledgements

We would like to express our gratitude to Mr. Christopher K. Mirabelli for his excellent technical assistance and to Dr. Jack V. Pierce, Hypertension-Endocrine Branch, National Heart and Lung Institute, NIH for human urinary kallikrein. This research was carried out under the auspices of the U.S. Department of Energy.

## References

- 1 Ossowski, L., Unkless, J.C., Tobia, A., Quigley, J.P., Rifkin, D.B. and Reich, E. (1973) *J. Exp. Med.* 137, 112—126
- 2 Unkless, J.C., Tobia, A., Ossowski, L., Quigley, J.P., Rifkin, D.B. and Reich, E. (1973) *J. Exp. Med.* 137, 85—111
- 3 Vassalli, J.-D., Hamilton, J. and Reich, E. (1977) *Cell* 11, 695—705
- 4 Wigler, M. and Weinstein, I.B. (1976) *Nature* 259, 232—233
- 5 Kaplan, A.P. and Austen, K.F. (1972) *J. Exp. Med.* 136, 1378—1393
- 6 Laake, K. and Venneröd, A.M. (1974) *Thromb. Res.* 4, 285—302
- 7 Mandle, R., Jr. and Kaplan, A.P. (1977) *J. Biol. Chem.* 252, 6097—6104
- 8 Kettner, C. and Shaw, E. (1977) in *Chemistry and Biology of Thrombin* (Lundblad, R.L., Mann, K.G. and Fenton, J.W., eds.), pp. 129—143, Ann Arbor Science Publishers, Inc., Ann Arbor, MI
- 9 Kettner, C. and Shaw, E. (1978) *Biochemistry* 17, 4778—4784
- 10 Sottrup-Jensen, L., Zajdel, M., Claeys, H., Petersen, T.E. and Magnusson, S. (1975) *Proc. Natl. Acad. Sci. U.S.A.* 72, 2577—2581
- 11 Coleman, P.L., Latham, H.G., Jr. and Shaw, E. (1976) *Methods Enzymol.* 45, 12—26
- 12 Green, G.D.J. and Shaw, E. (1979) *Anal. Biochem.* 93, 223—226
- 13 Shaw, E. (1970) *Physiol. Rev.* 50, 244—269
- 14 Kitz, R. and Wilson, I.B. (1962) *J. Biol. Chem.* 237, 3245—3249
- 15 Nustad, K., Gautvik, K.M. and Pierce, J.V. (1974) in *Chemistry and Biology of the Kallikrein-Kinin System in Health and Disease* (Pisano, J.J. and Austen, K.F., eds.), pp. 77—92 DHEW Publication No. (NIH) 76—791, Bethesda, MD
- 16 Robbins, K.C., Summaria, L. and Barlow, G.H. (1975) in *Proteases and Biological Control* (Reich, E., Rifkin, D.B. and Shaw, E., eds.), pp. 305—310, Cold Spring Harbor Press, Cold Spring Harbor, NY
- 17 Geratz, J.D. and Cheng, M.C.-F. (1975) *Thromb. Diath. Haemorrh.* 33, 230—243
- 18 Stürzebecher, J. and Markwardt, F. (1978) *Pharmazie* 33, 599—602
- 19 Landmann, H. and Markwardt, F. (1970) *Experientia*, 26, 145—147
- 20 Wong, S.-C., Green, G.D.J. and Shaw, E. (1978) *J. Med. Chem.*, 21, 456—459
- 21 Ong, E.B., Johnson, A.J. and Schoellmann, G. (1976) *Biochim. Biophys. Acta* 429, 252—257
- 22 Kettner, C., Springhorn, S., Shaw, E., Müller, W. and Fritz, H. (1978) *Hoppe-Seyler's Z. Physiol. Chem.* 359, 1183—1191
- 23 Coleman, P.L., Kettner, C. and Shaw, E. (1978) *Biochim. Biophys. Acta* 569, 41—51